

## STRUCTURAL INVESTIGATION OF SOME NATIVE BIODEGRADABLE PACKAGING MATERIALS

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**ABSTRACT.** The IR, Raman and NMR methods are used to investigate the structural effects of water and glycerol (plasticizers) on biodegradable packaging materials proceeding from native corn starch. The IR and Raman methods not give a clear response on the amorphous/crystalline content in the investigated starch samples because the origin of vibrational bands is in the same atomic groups of the major starch components (amylose, amylopectin). The nuclear magnetic relaxation data show that the amorphous/crystalline ratio depends not only by the amylose/amylopectin content but also of the mobility and the place of theirs polymer chain segments. The shapes of <sup>13</sup>C CP/MAS NMR spectra show clearly the presence of A, B, V-types crystalline structures and also amorphous phase which prevails with the increasing of plasticizers content and also in the extruded starch samples. The composition having starch/glycerol/water 68/17/15 mass % ratio was found to have a dominant amorphous character and very similar features with a commercial specimen (USA) used for the package. It was also found that this best package is extremely degraded after just one day under water absorption.

**Key words:** *starch, plasticizers, IR, Raman, NMR*

### INTRODUCTION

The native starch is used as the main component of biodegradable packaging materials and also in the textile, pharmacological and the food industry (Almeida et al., 2010; Liu et al., 2009; Luckachan and Pillai, 2011). It has a granular structure with an internal architecture characterized by concentric rings involving semi-crystalline shell (thickness 120-400 nm) separated by essentially amorphous regions (Wang et al., 2009). Chemically, the starch contains two glucose polymers amylose and amylopectin (Liu et al., 2006).

Amylose is essentially a linear structure of  $\alpha$ -1,4 linked glucose units, and amylopectin is a highly branched structure of short  $\alpha$ -1,4 chains linked by  $\alpha$ -1,6 bonds (Liu et al., 2009). Amylose and the branching points of amylopectin form the amorphous regions in the starch granules. Amylopectin is the main crystalline component in granular starch (Soesta and Vliegthartb, 1997). The physical-chemical properties of starch as viscosity, gelation power, adhesion etc., are strongly influenced by the amylose/amylopectin ratio (Almeida et al., 2010; Cereda et al., 2002).

The plasticizers molecule (water, glycerol) and other proceedings as annealing extrusion or enzymatic digestion can destabilize the crystalline regions and increase the amorphous domains (Brümme et al., 2002; Lopez-Rubio et al., 2008; Warren et al., 2016).

There are four types of crystalline structures in starch materials. From these the A-type is due to the junction of amylopectin short branches to backbone chain (lowest mobility) close to each other and appears in corn, wheat or rice cereals. In fruit and tuber starches prevails the B-type structure realized by amylopectin polymer chain segments located away from junctions in core branches region (more mobile). The V-type structure is formed by inclusion complexes between amylose and water, glycerol, alcohols molecules (plasticizers) stabilized by hydrogen bonds. There is also a C-type crystalline structure in leguminous plant, which is in fact a mixture of A and B-type structures (Cioica et al., 2013). Amorphous phase is due to mobile laterale branches of amylopectin and free chains of amylose.

The native starch materials used in the packaging process must to have a high content of amorphous phase and biodegradability at natural agents (Luckachan and Pillai, 2011). Besides the presence of plasticizers generally is also used the extrusion process what disrupts and transforms the semi-crystalline structure of starch granules in a homogeneous amorphous material with increased biodegradability (Mościcki et al., 2012; Nabar, 2006). The effect of the nature and content of the plasticizers (water, glycerol) on the native corn starch properties (crystalline/amorphous) and also on the degradation process after absorption of distilled water were investigated by vibrational (IR, Raman) and nuclear magnetic resonance (NMR relaxation,  $^{13}\text{C}$  CP/MAS NMR) spectroscopies.

## EXPERIMENTAL

The native corn starch used in this study was obtained from SC Amylon Sibiu, Romania, having water content of 10.76 %, a density of  $0.561 \text{ g/cm}^3$  and an amylose content of 21%. The glycerol used was purchased from SC Nordic Invest SRL Cluj-Napoca and had a concentration of 99.5% and a density of  $1.262 \text{ g/cm}^3$ . The water used was from the water supply system.

Two sets of samples with different starch / glycerol / water [%] ratios were prepared. First set of samples have the ratio starch / glycerol (= 4) constant and the added water content increased from 0 up to 15% (0; 2.5; 5; 7.5; 10; 12.5; 15). Samples from the second set have constant the ratio starch / water (= 5.7) and the glycerol content increases from 0 up to 16.5% (0; 2.5; 6; 9.5; 13; 16.5).

FT-IR/ATR spectra were recorded at room temperature on a conventional Equinox 55 (Bruker, Germany) spectrometer equipped with a DTGC detector, coupled with an ATR sampling device (Miracle, Pike Techn.). The resolution was of  $2\text{ cm}^{-1}$ . The ATR crystal is of ZnSe and we made 40 acquisition for obtaining IR spectra.

FT-Raman spectra were recorded with a resolution of  $4\text{ cm}^{-1}$  in a backscattering geometry with a Bruker FRA 106/S Raman accessory equipped with a nitrogen cooled Ge detector. The 1064nm Nd:YAG laser was used as excitation source, the laser power measured at the sample position was 300mW.

The  $^1\text{H}$  NMR relaxation measurements were obtained using the Bruker Minispec spectrometer with the 10 mm probe-head working at 19.688 MHz Larmor frequency. In all measurements the temperature was set at  $35\text{ }^\circ\text{C}$ . The CPMG (Car – Purcell – Meiboon – Gill) decays and saturation recovery curves were analyzed using the UPIN (Uniform – Penalty Inversion) algorithm, which perform a Laplace inversion of the measured data (Borgia et al., 1998). In this way the  $T_1$  (spin-lattice) and  $T_2$  (spin-spin) relaxation times distributions as function of samples composition were analyzed.

The  $^{13}\text{C}$  solid state cross-polarization magic-angle-spinning (CP/MAS) NMR spectra were recorded on a Bruker Avance III wide-bore spectrometer operating at 125 MHz Larmor frequency for  $^{13}\text{C}$ , using a 4 mm double resonance probe head. Standard  $^{13}\text{C}$  CP/MAS experiments were analyzed at room temperature, using a spinning frequency of 14 kHz and a  $^1\text{H}$   $90^\circ$  pulse of  $2.3\text{ }\mu\text{s}$ . The  $^{13}\text{C}$  CP/MAS spectra were calibrated using the  $^{13}\text{CH}_3$  line in TMS through an indirect procedure which uses the carboxyl NMR line in  $\alpha$ -glycine.

## RESULTS AND DISCUSSION

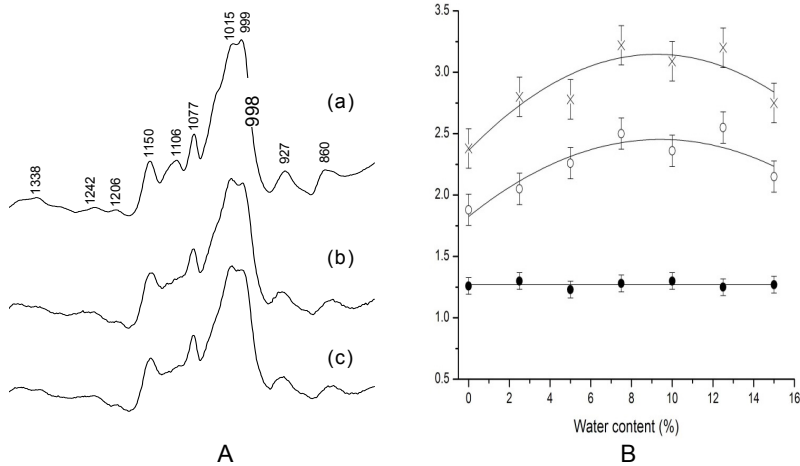
### *Vibrational spectra*

The characteristic IR spectra of starch samples with different amounts of plasticizer agents (water, glycerol) are situated in the  $2900\text{ cm}^{-1}$ ,  $1340\text{ cm}^{-1}$  and  $1000\text{ cm}^{-1}$  regions. These spectra contain also two bands at  $3300\text{ cm}^{-1}$  and  $1610\text{ cm}^{-1}$  due to the water stretching and bending vibrations (Morales et al., 2012; Iizuka and Aishima, 1999).

The absorptions situated at  $2930\text{ cm}^{-1}$  and  $2850\text{ cm}^{-1}$  are assigned to symmetric and asymmetric stretching vibrations of  $\text{CH}_2$  groups [16] and those from  $1350\text{ cm}^{-1}$  region to bending modes of O-C-H, C-C-H and C-O-H angles of amylose and amylopectin formations (Bellon–Maurel et al., 1995).

The peaks from the  $1150 - 900\text{ cm}^{-1}$  region are due to C-C and C-O stretching vibrations (Wilson and Belton, 1988).

According to Capron et al., 2007, the intense absorption from  $998\text{ cm}^{-1}$  may be assigned to crystalline domains whereas the band at  $1015\text{ cm}^{-1}$  reveals the amorphous contribution of plasticizers in starch samples (figure 1A).



**Fig. 1.** A. - The FT-IR spectra (in  $\text{cm}^{-1}$ ) of starch / glycerol / water = 68/17/15 mass% sample before (a) and after extrusion (b) and of commercial specimen – USA(c). B - The dependence of 998 / 2929(x), 1015/2929(o) and 998/1015(•) ratios band intensities versus water content

The evolution of the 998 / 2929 and 1015 / 2929 band intensity ratios versus water content (0 – 15 %) at a constant starch: glycerol ratio of 4 : 1 suggest a similar behavior (a parallelism) between increase and decrease of the crystalline / amorphous domains in the investigated starch samples with various plasticizers contents (figure 1B).

The constancy of 998 / 1015 ratio with the increase of water content suggests an absence of structural order at the FTIR observation range. A similar situation is also mentioned for the 1000 / 1022  $\text{cm}^{-1}$  ratio in the case of extruded starch and granular starch with water content under 20% (Capron et al., 2007).

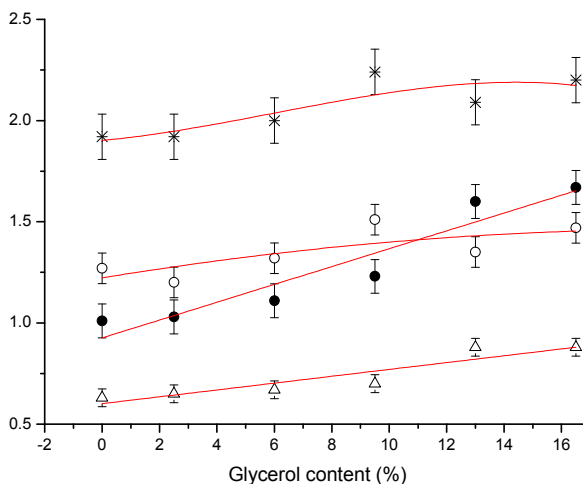
The band from 1015  $\text{cm}^{-1}$  becomes more intense than that situated at 998  $\text{cm}^{-1}$  in the extruded sample (figure 1A, b, c) being similar with a commercial specimen (USA). The extrusion process made with a co-rotating intermeshing twin-screw extruder ZK 25 (Collin, Germany), disrupts and transforms the semi-crystalline structure of starch granules in a homogeneous amorphous material which increase their biodegradability.

Raman spectra contain the two intense bands at  $\sim 2900 \text{ cm}^{-1}$  attributed to the symmetrical and antisymmetrical  $\text{CH}_2$  stretching vibrations (Almeida et al., 2010; Morales et al., 2012). Their intensity can be related to the amount of amylose and amylopectin content in the starch samples (Kizil et al., 2002).

The 1461  $\text{cm}^{-1}$  band corresponds to the superposition of CH,  $\text{CH}_2$  and C-O-H bending vibrations (Benzerdjeb et al., 2007) and those from 1380  $\text{cm}^{-1}$  and 1340  $\text{cm}^{-1}$  are due to C-O-H deformation modes and to the superposition of C-O stretching and C-O-H deformations, respectively (Almeida et al., 2010; Gussem et al., 2005).

The  $1200\text{ cm}^{-1}$  -  $1000\text{ cm}^{-1}$  region is characteristic of the C-O and C-C stretching and C-O-H deformation modes (Nikonenko et al., 2005; Yang and Zhang, 2009).

The bands observed at  $940\text{ cm}^{-1}$  and  $860\text{ cm}^{-1}$  are due to the superposition of bending C-O-C, C-O-H and stretching C-O vibrations from amylose  $\alpha$ -1,4 glycosidic linkage (Almeida et al., 2010) and also to the superposition of C-C-H and C-O-C deformations, respectively (Gussem et al., 2005).



**Fig. 2.** The dependence of 2900 / 1123 (\*), 1460 / 1340(●), 2900/477 (o) and 869 / 940 (Δ) ratios band intensities versus glycerol content and starch / water ratio constant (5.7). The errors bars are under 5% error limits

An other intense  $477\text{ cm}^{-1}$  Raman band is due to the superposition of bending C-C-C and twisting C-O vibrations and is used as a marker to identify the presence of starch in different samples.

The glycerol ( $\text{OH} \cdot \text{CH}_2 \cdot \text{CH}(\text{OH}) \cdot \text{CH}_2 \cdot \text{OH}$ ) stretching  $\nu_s, \nu_{as} - \text{CH}_2$  vibrations, the in plane  $\delta - \text{CH}_2$  deformations and CC stretching vibrations contribute to the following starch bands:  $2900\text{ cm}^{-1}$ ,  $1461\text{ cm}^{-1}$  and  $860\text{ cm}^{-1}$ , respectively.

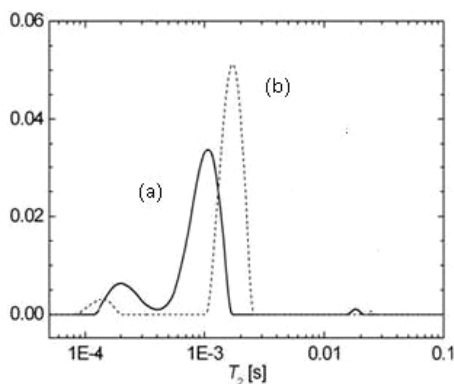
An easy increase of the 2900/1123 (\*), 2900/477 (o) and 868/940 (Δ) ratios band intensities may be observed for the samples, with starch / water ratio of (5.7) and the added glycerol increasing up to 16.5% (figure 2). These suggest that the amylose / amylopectin ratio remains practically unchanged and plasticizing – antiplasticizing effect depends on the levels of realized hydrogen bondings which lead to decrease - increase the rigidity of starch components.

The more pronounced increase of the 1460/1340 (●) band ratio is due to the glycerol contribution at  $1461\text{ cm}^{-1}$  band by the in plane bending  $\text{CH}_2$  vibrations from the inclusion complex realized between amylose and glycerol molecules by hydrogen bonds (V – type structure).

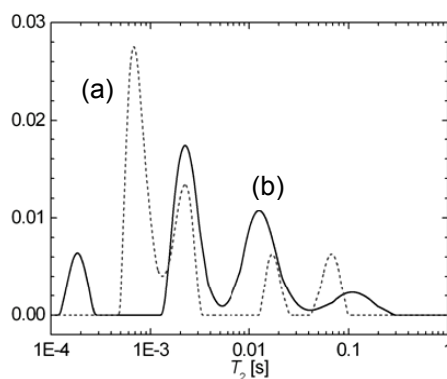
### NMR data

The nuclear magnetic relaxation method (Cioica et al., 2013) offer information's on the structure of monomer units by direct spin–spin interactions ( $T_2$ ) and also on the supramolecular organization by spin diffusion, spin–lattice relaxation ( $T_1$ ).

A number of three peaks, two intense at  $T_2 \sim 0.2$  ms and  $T_2 \sim 1$  ms and also a small peak at  $T_2 \sim 20$  ms are observed for pure starch and for a mixture of starch with water (figure 3).



**Fig. 3.** The spin-spin relaxation times  $T_2$  distribution of starch (a) and 85% starch + 15% water (b)



**Fig. 4.** The spin-spin relaxation times  $T_2$  distribution of 80% starch + 20% glycerol (a) and 68% starch + 17% glycerol + 15% water (b)

The peak from  $T_2 \sim 0.2$  ms may be associated with starch polymer chain segments with the lowest mobility (A – type structure) due to the junction of one end of amylopectin branches to backbone chain. The second component with  $T_2 \sim 1$ ms may be associated with more mobile amylose and/or amylopectin polymer chain segments located away from junctions in core branches region (B – type centers). The minor  $T_2$  component located at  $\sim 20$  ms presents a dynamic liquid-like behavior and can be associated with the mobile lateral branches of the amylopectin microstructure or free chains of amylose microstructure (amorphous). The small integral area of this peak suggests that the percentage of lateral branches is small, leading to the idea that the starch molecules are closely packed (crystalline).

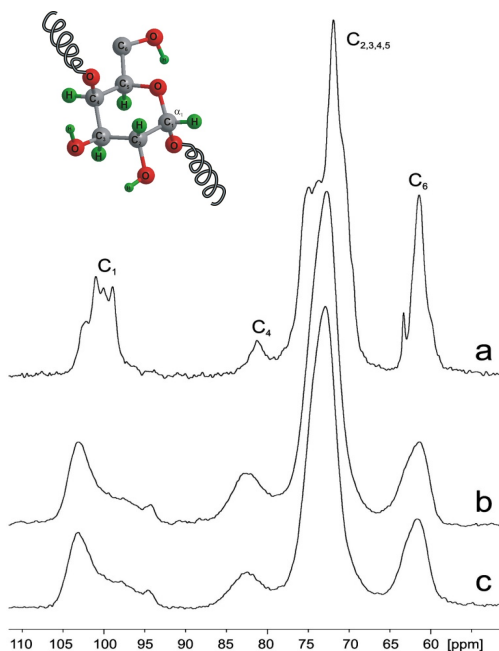
By adding 15 % of water one can observe a displacement of semi-mobile ( $T_2 \sim 1$ ms) and mobile ( $T_2 \sim 20$ ms) peaks to higher  $T_2$  values and a decrease of the rigid component ( $T_2 \sim 0.2$  ms). Thus the additional water content lead to a larger mobility of amylopectin polymer chain segments in the core packed branches.

The mixture between starch with glycerol (Fig.4a) leads to a distribution characterized by four peaks. The significant dynamic components remain in the semi-mobile region (peaks centered at  $T_2 \sim 0.6$  ms and 2.5 ms) and the other two peaks situated at larger  $T_2$  values of  $\sim 20$  ms and  $\sim 70$  ms suggest that glycerol

increases the number of lateral branches and may create mobile free ends in amylose and amylopectin polymer chain segments finally leading to an increase of mobility of the starch polymer chains (amorphous).

An increase in intensity of the last three major components can be observed in the  $T_2$  distribution for the mixture of starch with glycerol and water (Fig.4b). This can be interpreted as a transformation of amylopectin microstructure where the lateral branches are moved away from the core branches with the creation of more mobile free ends (amorphous). However, it should be noted that the presence of additional water molecules can maintain a rigid dynamic component (crystalline) of the starch polymer chain segments (peak located at  $\sim 0.2$  ms) corresponding to amylopectin segments around junctions.

The  $^{13}\text{C}$  CP/MAS NMR spectrum (figure 5a) suggest the presence of both crystalline and amorphous phases in the case of the 68% starch/ 17%glycerol/ 15%water sample as shown also by the NMR relaxation data (figure 4b).



**Fig. 5.** The  $^{13}\text{C}$  CP/MAS NMR spectra of starch / glycerol / water [mass %] formula with 68/17/15 ratio, a – before and b – after extrusion and c – commercial USA extruded product, recorded at room temperature and 14 kHz spinning speed

Before extrusion, the A, B and V- crystalline structures are present (figure 5a). In the region of the glucose unit C1 NMR resonance there is an overlap of the *triplet* at  $\delta = 101$  ppm (A phase), the *doublet* at  $\delta = 101$  ppm (B structure) and a *singlet* at  $\delta = 104$  ppm (V type structure).

The strong C 2,3,4,5 *quadruplet* from  $\delta = 73$  ppm is characteristic of A – type phase.

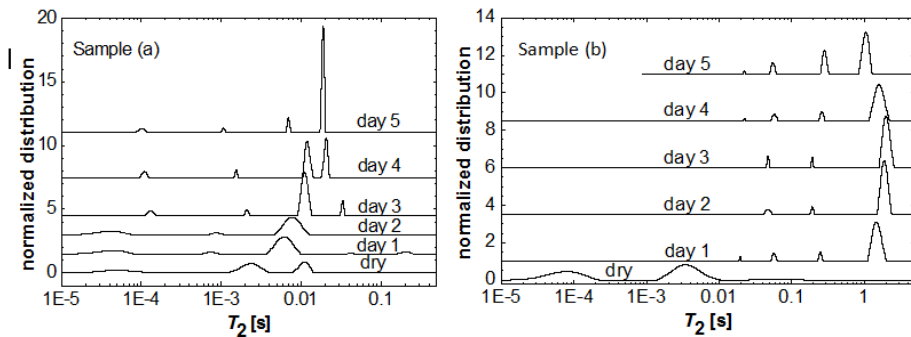
The intense C6 absorption ( $\delta = 62$  ppm) is due to contributions from A, B, and V – types crystalline phases (Wang et al., 2009; Therien and Zhu, 2009). All these absorptions are situated over the weak and broad NMR resonances at  $\delta = 104$  ppm (C1),  $\delta = 83$  ppm (C4) and  $\delta = 61$  ppm (C6) characteristic of the amorphous phase as in the extruded samples (b, c).

The sharp  $^{13}\text{C}$  line at  $\delta = 64$  ppm belongs to inclusion glycerol – amylose complex stabilized by hydrogen bonds.

### Degradation process

Environmental degradation of biopolymers may be defined as the scission of chemical bonds in the polymer backbone by the attack of water to form oligomers and finally monomers (Shah et al., 2008). The degradation of starch – based packaging required unstable and hydrolysable linkages where chemical, biological or photochemical reactions can take place with the disintegration into their monomers (Wollendorfer and Bader, 1998).

By following the absorbed water process during five days we found that the product with the high starch content (78/19.5/2.5 starch/glycerol/water – (a)) is most resistant and start to be degraded after only five days (figure 6a).



**Fig. 6.** The normalized  $T_2$  relaxation times distributions of packages samples (a) and (b) for 5 days water absorption compared with dry sample

A semi-mobile phase ( $T_2 \sim 2.5$  ms and  $\sim 11$  ms) and also a rigid region ( $T_2 \sim 50$   $\mu\text{s}$ ) due to the junctions of lateral branches with the amylopectin polymer backbone can be observed for dry sample (a). After one day the two mobile peaks collapses to a unique peak located at  $T_2 \sim 6.3$  ms (figure 6).

Dramatically changes can be observed starting with day 3 when that the  $T_2$  peaks becomes more narrow indicating a reduced distribution compared dray sample or with not so degraded sample (first two days). Their shifts to larger  $T_2$  values compared with the values measured in day 2, suggest the presence of most mobile components.



The product with lower starch content (68/17/15 starch/glycerol/water – (b)) is degraded after one day becoming a colloidal aspect. The dry sample (b) is characterized by two relative rigid components ( $T_2 \leq 5$  ms). The amorphous phases prevail in this sample ( $T_2 > 10$  ms) after one day (figure 6).

With day 4 a precipitation validated by the decrease in intensity and shifted at lower values of the main peak (more mobile  $T_2 \sim 1$  s) can be observed for sample (b). A similar behavior was also observed for commercial USA specimen.

## CONCLUSION

The IR and Raman methods not give a clear response on the amorphous or crystalline content in starch samples because the vibrational bands have the origin in the same atomic groups of the major starch components.

The relaxation NMR spectroscopy shows that the amorphous / crystalline ratio depends on amylose / amylopectin mobility and also by the place of their polymer chain segments. The distributions of spin – spin ( $T_2$ ) and spin – lattice ( $T_1$ ) relaxation times values and their shift toward higher values show that the increasing of water and glycerol content in the starch package samples (68/17/15) lead to the more mobile amylose and amylopectin polymer chain segments and the prevalence of amorphous phase.

$^{13}\text{C}$  CP/MAS NMR spectra show also that in the extruded samples the amorphous phase becomes dominant.

The most amorphous package with starch/glycerol/water of 68/17/15 mass% ratio is extremely degraded after just one day by the water absorption, forming a colloid substance. The package with a large content of starch (78/19.5/2.5) is most resistant, starting to be decomposed after five days.

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