

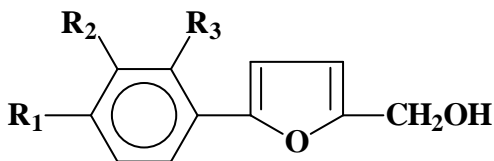
MASS SPECTROMETRY OF SOME NEW 2-HYDROXYMETHYL-5-PHENYL-FURANS OBTAINED THROUGH CELL CATALYSIS

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ABSTRACT. Eight 2-hydroxymethyl-5-phenyl-furans were obtained from the corresponding aldehydes through reaction with *Saccharomyces cerevisiae*. The fragmentation scheme showed the $M^+ - 17$ peaks for every system studied and characteristic features for the heterocyclic moiety of the alcohols. Some details are discussed within the fragmentation scheme for each compounds.

INTRODUCTION

As a part of our researches in the furan series we have prepared eight new from the corresponding aldehydes using *Sacharomices cerevisiae* as reducing agent [6].



a: R₁ = F; R₂ = H; R₃ = H
b: R₁ = Cl; R₂ = H; R₃ = H
c: R₁ = Br; R₂ = H; R₃ = H
d: R₁ = I; R₂ = H; R₃ = H

e: R₁ = H; R₂ = Br; R₃ = H
f: R₁ = H; R₂ = H; R₃ = Cl
g: R₁ = H; R₂ = H; R₃ = Br
h: R₁ = H; R₂ = H; R₃ = I

Scheme 1

This paper deals with the mass spectra of these compounds. We considered it of interest to find out to what extent the characteristic fragmentation modes of furans[1-5], are also presents in the case of a-h and what is the influence of the hydroximethyl group in position 2 on the furan ring upon the fragmentation process.

Mass spectral data m/e (I%):

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- a: 51(10,6), 57(8,6), 75(15,5), 83(9,6), 91(10,5), 95(19,5), 107(10,8), 109(10,8), 118(11,3), 120(6,4), 123(34,8), 133(48,3), 134(11,2), 135(10,1), 145(12,7), 146(23,7), 147(19,6), 163(9,1), 164(8,7), 175(99), 176(13), 187(3), 188(7,3), 192(100), 191(13)
- b: 29(13,4), 39(12), 46(21,3), 63(31,4), 75(41,3), 87(26,3), 91(11), 111(33), 112(6), 113(16,7), 114(19,8), 115(41,6), 116(22,2), 117(11,6), 127(33,8), 128(41,6), 139(39,7), 140(9,1), 141(9,3), 142(3), 149(43,6), 150(6,3), 151(22,8), 161(2), 162(8,3), 163(7,1), 164(5), 165(4), 178(6), 179(8), 180(10,1), 181(6,3), 182(4,1), 191(74,5), 192(13,6), 193(26,3), 194(5,1), 205(10,1), 206(23,1), 208(100), 209(25,3), 210(33,5), 211(6,1)
- c: 29(10), 50(11), 63(18,2), 78(13), 88(16), 91(13), 109(33), 110(43,6), 111(43,6), 112(43,6), 113(53), 114(41,6), 115(21,6), 127(30), 128(57,2), 129(59,6), 144(27), 145(13), 155(14), 157(14), 183(32), 185(32), 193(12), 195(12), 224(13), 226(13), 235(62), 236(20), 236(62), 237(20), 250(8), 251(13,6), 252(100), 253(31,3), 254(100), 255(10)
- d: 63(3), 74(2), 88(4), 91(11,2), 115(13), 135(6,3), 149(2), 171(2,6), 203(6,3), 241(21), 270(6), 283(81), 298(60), 299(30), 300(100), 301(10)
- e: 29(10), 50(11), 63(18,2), 78(13), 88(16), 91(13), 109(33), 110(43,6), 111(43,6), 112(43,6), 113(53), 114(41,6), 115(21,6), 127(30), 128(57,2), 129(59,6), 144(27), 145(13), 155(14), 157(14), 183(32), 185(32), 193(12), 195(12), 224(13), 226(13), 235(62), 236(20), 236(62), 237(20), 250(8), 251(13,6), 252(100), 253(31,3), 254(100), 255(10)
- f: 29(13,4), 39(12), 46(21,3), 63(31,4), 75(41,3), 87(26,3), 91(11), 111(33), 112(6), 113(16,7), 114(19,8), 115(41,6), 116(22,2), 117(11,6), 127(33,8), 128(41,6), 139(39,7), 140(9,1), 141(9,3), 142(3), 149(43,6), 150(6,3), 151(22,8), 161(2), 162(8,3), 163(7,1), 164(5), 165(4), 178(6), 179(8), 180(10,1), 181(6,3), 182(4,1), 191(74,5), 192(13,6), 193(26,3), 194(5,1), 205(10,1), 206(23,1), 208(100), 209(25,3), 210(33,5), 211(6,1)
- g: 29(10), 50(11), 63(18,2), 78(13), 88(16), 91(13), 109(33), 110(43,6), 111(43,6), 112(43,6), 113(53), 114(41,6), 115(21,6), 127(30), 128(57,2), 129(59,6), 144(27), 145(13), 155(14), 157(14), 183(32), 185(32), 193(12), 195(12), 224(13), 226(13), 235(62), 236(20), 236(62), 237(20), 250(8), 251(13,6), 252(100), 253(31,3), 254(100), 255(10)
- h: 63(3), 74(2), 88(4), 91(11,2), 115(13), 135(6,3), 149(2), 171(2,6), 203(6,3), 241(21), 270(6), 283(81), 298(60), 299(30), 300(100), 301(10)

RESULTS AND DISCUSSION

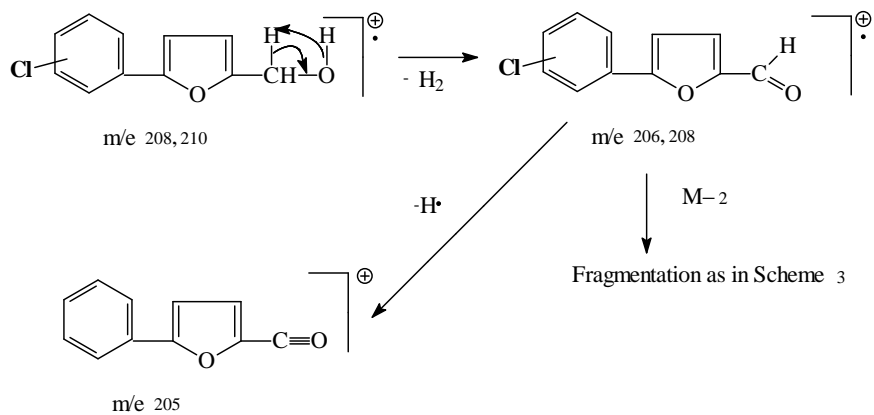
In order to simplify the interpretation of the mass spectra of our compounds we recorded also the spectrum of 5-phenyl-furyl-2-carboxaldehydes substituted with halogenes at the phenylic ring as model.

MASS SPECTROMETRY OF SOME HYDROXYMETHYL-PHENYL-FURANS

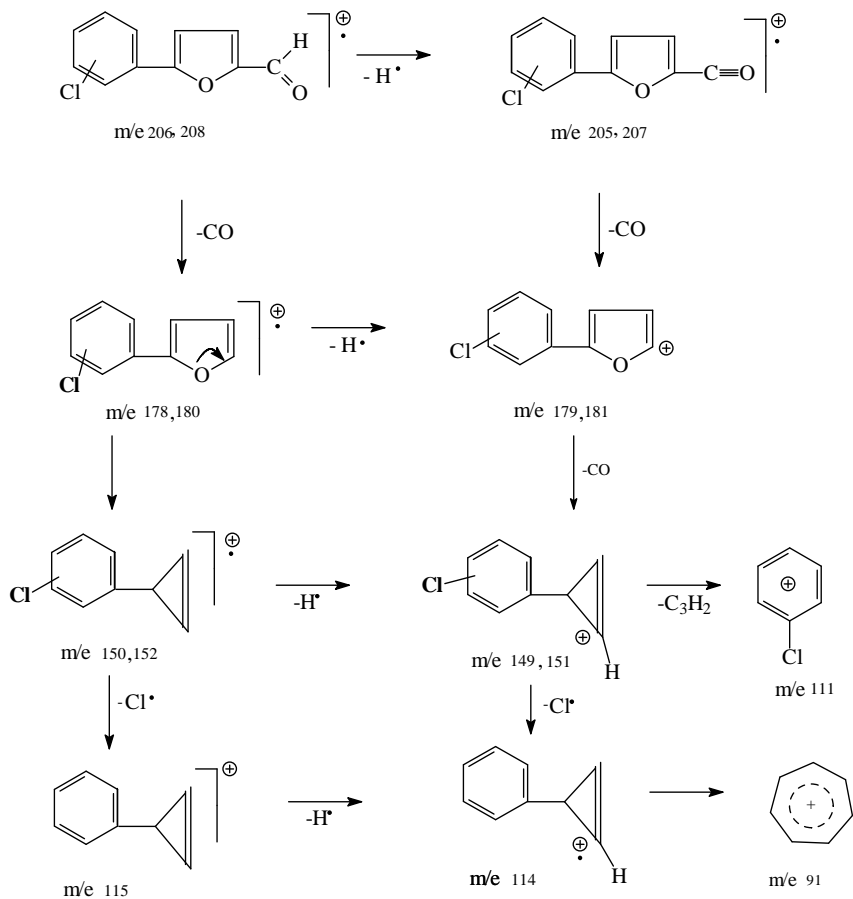
In case of bioproducts we observed a fragmentation characteristic for benzylic alcohols. The molecular ion is the base peak $M - 1$, $M - 2$, $M - 3$ are observed too. Slow intensity $M+1$ peak is appeared.

For exemplification we presented the fragmentation mode of chloroderivatives.

The fragments presents in Scheme 3 are also appeared in the mass spectras of the corresponding aldehydes too.

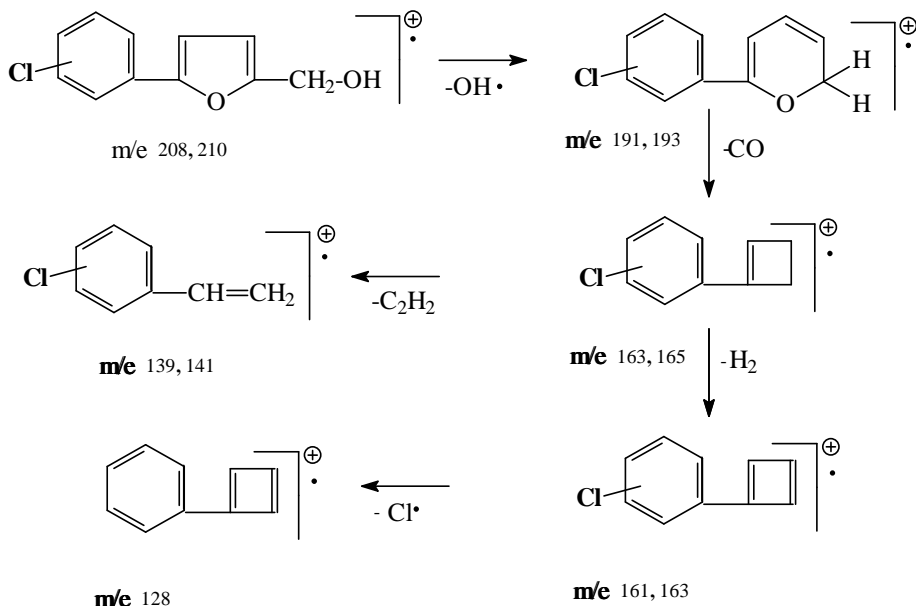


Scheme 2



Scheme 3

An other way for the cleavage presented in Scheme 4 is with elimination of an $\text{-OH}\cdot$ radical. We supposed that the $M - 17$ suffered a ring rearrangement which cleaves with elimination of -CO and gives a cyclobutadienylic ring as in case of hydroxymethyl-furyl benzothiazoles [7].



Each substances presents the same pattern of the fragmentation. In case of c, e, g its observed the presence of two bromine isotope (79,81).

The substance presents two molecular peaks ($M = 252, 254$) with the same abundance and two characteristic fragments for the benzylic alcohols, $M-17$ (225, 227). For substances b and f the two chloro isotope (35, 37) are present, two molecular peaks (208, 210) and two characteristic fragments, $M-17$ (191, 193) are appeared.

Molecular peaks for (a-h) are clear with an relative abundance of 100%.

EXPERIMENTAL

The purity of the substances was checked by TLC on silica gel. The mass spectra were recorded using a Double focusing VARIAN MATT 311 Spectrometer with an electronic impact at 70 eV and 300 mA. The source temperature was 150-200°C.

REFERENCES

1. R.I. Reed, W.K. Reid, *J.Chem.Soc.*, 1963, 5933.
2. H. Budzikiewicz, C. Djerassi, D.H. Williams, *Mass Spectrometry of Organic Compounds*, Holden-Day Inc., 1968, 615.
3. R. Grigg, M.W. Sargent, D.H. Williams, J.A. Knight, *Tetrahedron*, 1965, **21**, 3441.
4. K. Heyns, R. Stute, M. Scharmann, *Tetrahedron*, 1966, **22**, 2223.
5. C. Mercier, *Bull.Soc.Chim.France*, 1969, 145.
6. F.D. Irimie, Cs. Paizs, R. Silaghi-Dumitrescu, C. Majdik, Fr. Joo, M. Toșa, *Synth.Commun.*, 1999, (in press).
7. F.D. Irimie, Cs. Paizs, V. Miclăuș, C. Afloaroaei, M. Toșa, G. Damian, *Balkan Phys. Lett*, 1997, **5**, 227.