

## SYNTHESIS OF *CIS* - 7,8 – EPOXY - OCTADECANE, SPECIES - SPECIFIC COMPONENT OF THE SEX PHEROMONE OF NUN MOTH *LYMANTRIA MONACHA* (LEPIDOPTERA, LIMANTRIIDAE)

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**ABSTRACT.** The synthesis of *cis*-7,8-epoxy-octadecane, species-specific component of the sex pheromone of nun moth *Lymantria monacha* was based on a  $C_{10}+C_2=C_{12}$  and  $C_{12}+C_6=C_{18}$  coupling scheme, by acetylenic route, starting from 1-bromo-decane. The second coupling reaction took place between di-*n*-dodecyne mercury and 1-bromo-hexane in a transmetallation reaction.

**Keywords:** nun moth, *Lymantria monacha*, *cis*-7,8-epoxy-octadecane

### Introduction

Nun moth (NM), *Lymantria monacha*, in Europe and gypsy moth (GM) *Lymantria dispar*, in Europe and North America are most important defoliators of coniferous and deciduous forests, respectively.

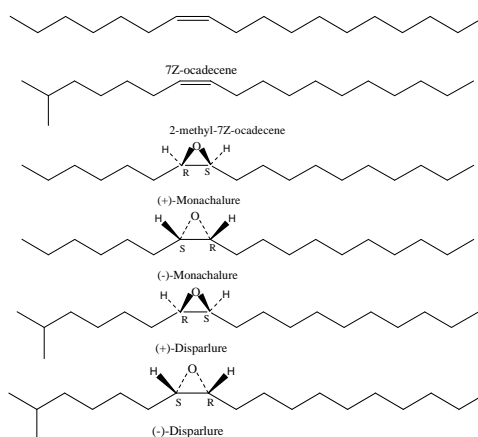


Figure 1

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For sexual communication GM and NM females produce the pheromone disparlure; NM and GM males respond to the (+) - disparlure enantiomer [(7R, 8S)-cis-7,8-epoxy-2-methyloctadecane] as a single attractive but unspecific pheromone component [1-29].

Three epoxies [(-) - disparlure, (+) - and – (-) - monachalure] and two hydrocarbon volatile components [2-methyl- Z7-octadecene and Z7-octadecene] [30] synergistically prevented cross-attraction of coseasonal GM males and imparted specificity to NM sexual communication. Fig.1

In the literature are not mentioned the synthesis of cis-7,8-epoxy-octadecane, species-specific component of the sex pheromone of nun moth *Lymantria monacha*.

The paper presents a route for the synthesis of cis-7,8-epoxy-octadecane.

### Results and Discussion

For the synthesis of cis-7,8-epoxy-octadecane (1) we explored the pathway outlined in Scheme 1, based on a  $C_{10}+C_2=C_{12}$  and  $C_{12}+C_6=C_{18}$  coupling scheme, by acetylenic route, starting from 1-bromo-decane.

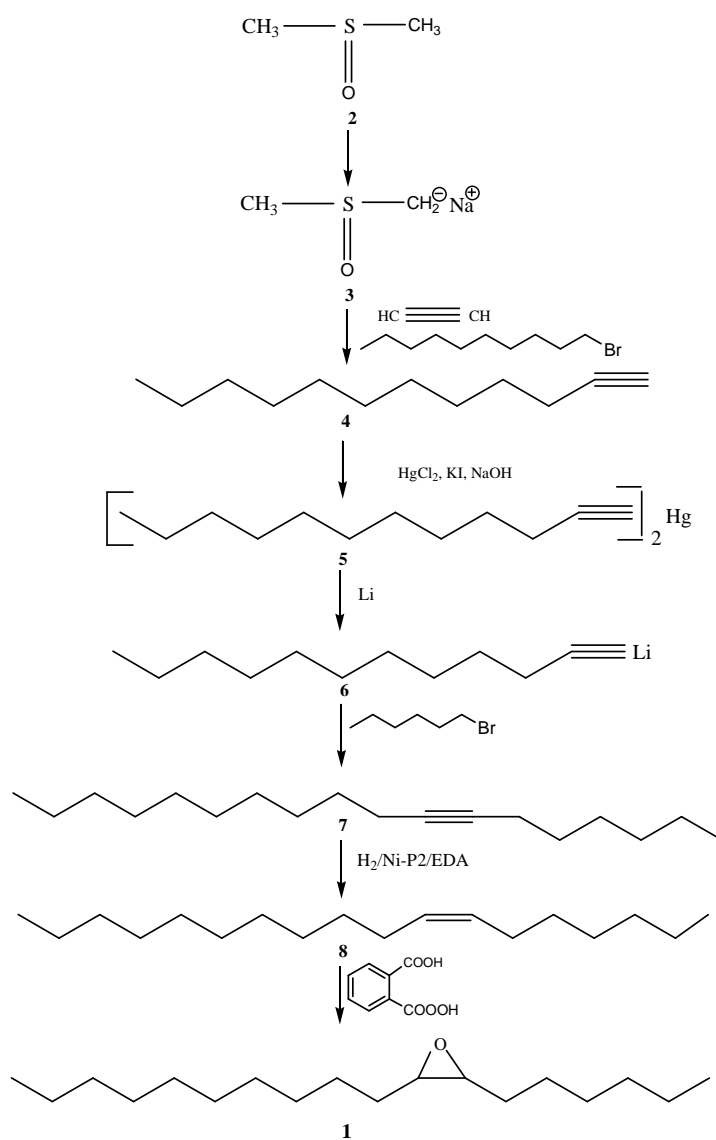
In our variant, the first coupling reaction was effected in order to obtain 1-dodecyne (4) by adding 1-bromo-decane to monosodiate acetylene, obtained in situ from acetylene in DMSO with sodium hydride. 1-Dodecyne (4) with mercuric chloride in an aqueous solution containing potassium iodide and sodium hydroxide (Nessler's reagent [31]) was quantitatively precipitated as di-n-dodecynyl mercury (5).

The key step in our acetylenic route consisted in transmetallation of di-n-dodecynyl mercury (5), which was directly lithiated by heating with lithium metal in diglyme and then alkylated with 1-bromo-hexane obtaining 7-octadecyne (7).

Z-7-Octadecene (8) was prepared from 7-octadecyne (7) by hydrogenation using Ni-P2/EDA catalyst, obtained by reducing nickel acetate with sodium borohydride and subsequent poisoning with ethylenediamine [32]. Using NiP-2 catalyst, the hydrogenation takes place quantitatively and with a very high degree of stereoselectivity.

To prepare cis-7,8-epoxy-octadecane (1) the olefine 8 was epoxidated with monophtalic acid.

SYNTHESIS OF *C/S* - 7,8 – EPOXY - OCTADECANE, SPECIES - SPECIFIC COMPONENT ...



**Scheme 1**

**Experimental**

GC-MS analyses were performed on a Hewlett Packard GL-MSD 5890-5972 instrument, using a HP-5MS 30mX0.25mmX0.25μm capillary column. A Perkin Elmer Spectrometer Model 700 was used for IR spectra.

#### **1-Dodecyne (4)**

Aparatus: 1L reactor provided with a thermometer dipping into the liquid, a dropping funnel, a stirrer and a 20-30 cm long tube for preventing any splashing out of the reaction mixture.

All operations took place in inert atmosphere and under vigorous stirring.

4,41g (92 mmoles) of sodium hydride were added to 56 mL of dry DMSO. The temperature was risen up to 65<sup>0</sup>-68<sup>0</sup>. The conversion was finished when no more hydrogen was evolved.

Acetylene (freed from acetone) was introduced for 15 min. at a rate of about 2L/min with external cooling in order to moderate the reaction.

During this process the temperature is maintained between 20<sup>0</sup> and 30<sup>0</sup>C.

28 mL of DMSO were added and afterwards, between 20<sup>0</sup>-30<sup>0</sup>, 8,17 g (37 mmoles) of 1-bromo-decane with external cooling.

The reaction was led to completion by stirring for 3 hours at room temperature. The reaction mixture was poured into 100 mL of ice-water and extracted with four portions of 100 mL petroleum ether.

The combined extract were washed with brine and dried over MgSO<sub>4</sub>. A satisfactory purity could be achieved by vacuum distillation collecting the fraction at 92<sup>0</sup>-94<sup>0</sup>C/28 mmHg; yield: 6,14g (79%)

*IR spectrum (film, cm<sup>-1</sup>):* 2160 s (-C≡C), 3280 vs (≡CH).

*Mass spectrum (m/z; %):* 151(<1), 137(<1), 123(1), 109(10), 95(40), 81(100), 67(77), 55(53), 41(85), 29(37).

#### **Di-n-dodecynyl mercury (5)**

18,4 g (110 mmoles) of 1-dodecyne (4) dissolved in 500 mL ethanol was added to a vigorous stirred solution of Nessler's reagent prepared according to the standard procedure [31].

The product 5 was obtained in yield of 93%, m.p.= 86<sup>0</sup>-87<sup>0</sup>.

#### **7-Octadecyne (7)**

16,33 g (30 mmoles) of di-n-dodecynyl mercury (5) in 84 mL diglyme was treated with 0,44 g of lithium under inert atmosphere at 100-110<sup>0</sup> for 2 hours.

11g (66 mmoles) of 1-bromo-hexane was added dropwise then the temperature raised to 120<sup>0</sup>-130<sup>0</sup> under stirring for 6 hours. The mixture was left overnight, decanted, diluted with water and extracted with ethyl ether. The ethereal solution was washed to neutral and dried over MgSO<sub>4</sub>.

After the removal of the solvent 12,1g (75%) of 7 were obtained, GC purity: 92%.

*Mass spectrum (m/z; %):* 250(<1), 235(<1), 221(<1), 207(<1), 180(1), 165(4), 152(2), 123(12), 109(29.9), 95(44.5), 81(53), 55(76.04), 67(54), 40(100), 29(24).

#### **Z-7-Octadecene (8)**

7-Octadecyne (7) was hydrogenated at Z-7-octadecene (8) in the presence of the NiP2/EDA catalyst, prepared in situ according to the standard procedure [32]. Z-7-octadecene (8) was obtained in yield of 93% , GC purity: 95%.

*Mass spectrum (m/z, %):* 252(10), 224(1), 196(<1), 182(<1), 168(1), 158(1.9), 125(11), 111(31.2), 97(59.8), 83(65.5), 69(73.4), 55(100), 41(99.5), 29(44.9).

**Z-7,8-Epoxy-octadecane (1)**

7,56g (30 mmoles) of Z-7-octadecene (8) was treated with 68 mL of ethereal solution of monoperoxyphthalic acid (1 mL containing 93 mg of peracid) and kept at room temperature for 24 hours. The mixture was then filtered, the Synthesis of *cis* - 7,8 - epoxy - octadecane filtrate washed successively with 10% NaHCO<sub>3</sub>, diluted NaOH and water. After drying over MgSO<sub>4</sub>, the solvent was removed and the residuum was **1**. Yield: 95%, GC purity: 95%.

Mass spectrum (*m/z*, %): M=268(<1), 252(<1), 225(<1), 211(<1), 197(<1), 183(12), 152(2.5), 141(<1), 127(15), 97(55.9), 83(40.9), 69(59), 55(100), 41(84), 29(45.5).

**REFERENCES**

1. B.A. Bierl, M. Beroza, C.W. Collier, *Science*, **1970**, 170, 87
2. B.A. Bierl, M. Beroza, C.W. Collier, *J. Econ. Entomol.*, **1972**, 65, 659
3. H.J. Bestmann, O. Vostrowsky, *Tetrahedron Letters*, **1974**, 207
4. H.J. Bestmann, O. Vostrowsky, W. Stransky, *Chem. Ber.*, **1976**, 109, 3375
5. T.H. Chan, E. Chang, *J. Org. Chem.*, **1974**, 39, 3264
6. S. Iwaki, S. Marumo, T. Saito, M. Yamada, K. Katagiri, *J. Am. Chem. Soc.*, **1974**, 96, 7842
7. K. Mori, T. Takigawa, M. Matsui, *Tetrahedron Letters*, **1976**, 3953
8. J.H. Margraf, S.I. Lussking, E.C. McDonald, B.D. Volpp, *J. Chem. Ecol.*, **1983**, 9211
9. H.C. Brown, D. Basavaian, *Synthesis*, **1983**, 283
10. K. Koumaglo, H.T. Chan, *Tetrahedron Letters*, **1984**, 25, 717
11. S. Tsuboi, H. Fukutani, A. Takeda, K. Kawazoe, S. Sato, *Bull. Chem. Soc. Japan*, **1987**, 60, 2475
12. G.Q. Lin, B.C. Wu, L.Y. Liu, X.Q. Xian, W.S. Zhou, *Acta Chem. Sinica*, **1984**, 74
13. S. Pikul, M. Kozłowska, J. Jurezak, *Tetrahedron Letters*, **1987**, 28, 2627
14. Y. Masaki, Y. Serizawa, K. Nagata, H. Oda, H. Nagashima, K. Kaji, *Tetrahedron Letters*, **1986**, 27, 231
15. O. Jr. Achmatowich, A. Sadownik, R. Bielski, *Polish. J. Chem.*, **1985**, 59, 553
16. V.B. Jigajimmi, R.H. Wightman, *Carbohydrate Res.*, **1986**, 147, 145
17. A.G. Tolstikov, N.V. Kharhalina, V.N. Odinikov, *Zhur. Org. Khim.*, **1987**, 23, 2469 and *Zhur. Org. Khim.*, **1989**, 25, 296
18. R.E. Rossiter, T. Katsuki, K.B. Sharpless, *J. Am. Chem. Soc.*, **1981**, 103, 464
19. K. Mori, T. Ebata, *Tetrahedron Letters*, **1981**, 22, 4281
20. K. Mori, T. Ebata, *Tetrahedron Letters*, **1986**, 42, 3471
21. S. Marczak, M. Masnyk, J. Wicha, *Tetrahedron Letters*, **1989**, 30, 2485
22. G.Q. Lin, Y. Jiang, W.S. Zhou, *Acta Chem. Sinica*, **1985**, 257
23. T. Satoh, T. Oohara, Y. Ueda, K. Yamakawa, *Tetrahedron Letters*, **1985**, 29, 313
24. T. Satoh, T. Oohara, Y. Ueda, K. Yamakawa, *J. Org. Chem.*, **1989**, 54, 3130
25. S. Tsuboi, H. Futurani, M. Utaka, A. Takeda, *Tetrahedron Letters*, **1987**, 28, 2709

26. Bianchi, W. Cabri, P. Cesti, F. Francalanci, F. Rama, *Tetrahedron Letters*, **1988**, 29, 2455
27. Otto, F. Stein, C.A. van der Willigen, *Agric. Ecosyst. Environ.*, **1988**, 21, 121
28. T. Sato, T. Itoh, T. Fujisawa, *Tetrahedron Letters*, **1987**, 28, 5677
29. Y. Ko. Soo, *Tetrahedron Letters*, **1994**, 35, 3601
30. G. Gries, R. Gries, *Naturwissenschaften*, **1996**, 83, 382-385
31. E. Muller, J.K. Johnson, W.L. McEwen, *J. Amer. Chem. Soc.*, **1926**, 48, 469
32. C.A. Brown and V.K. Ahnya, *Chem. Commun.*, **1973**, 553