

## CORRELATING STUDY OF NEW MOLECULAR GRAPH DESCRIPTORS

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**ABSTRACT.** A correlating study, using as independent variables the topological indices newly developed within the design of the super-index Cluj-Niš *CJN*, is performed on the set of octane isomers. Among the modeled properties, the boiling point BP, entropy S and total surface area TSA, gave best scores, in mono- to three-variable regressions, with respect to our novel descriptors. The most important result was the monovariate description of octane boiling points, which is the best result so far published in literature.

**Keywords:** Octanes, topological indices, QSPR.

## INTRODUCTION

Alkanes represent an interesting class of compounds as a starting point for the application of molecular modeling procedures. Many properties of alkanes vary function of molecular mass or branching, and alkanes can be described by using a single type of (carbon) atom. There are properties well accounted by a single molecular descriptor, e.g., octane number MON, entropy S, molar volume MV, molar refraction MR, etc. Other properties, such as boiling point BP, heat of vaporization HV, total surface area TSA, partition coefficient LogP, density DENS, critical temperature CT, critical pressure CP, and heat of formation DHF, are notable exceptions, being not well modeled by a single parameter [1,2].

The purpose of the present report is to evaluate the relative performances of new topological indices developed within the design of super-index Cluj-Niš *CJN* [3] in relating the hydrocarbon molecular structures to a set of physical properties.

## LAYER MATRICES

Layer matrices have been proposed in connection with sequences of metrics: *DDS* (Distance Degree Sequence), *PDS* (Path Degree Sequence), and *WS* (Walk Sequence) [4-8]. They are built up on layer partitions in a graph.

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Let  $G(v)_k$  be the  $k^{th}$  layer of vertices  $v$  lying at distance  $k$ , in the partition  $G(i)$ :

$$G(v)_k = \{v \mid d_{i,v} = k\} \quad (1)$$

$$G(i) = \{ G(v)_k ; k \in [0, 1, \dots, ecc_i] \} \quad (2)$$

with  $ecc_i$  being the *eccentricity* of  $i$ . The entries in the layer matrix (of a vertex property) **LM**, are defined as

$$[LM]_{i,k} = \sum_{v \mid d_{i,v}=k} p_v \quad (3)$$

The layer matrix is a collection of the above defined entries:

$$LM(G) = \{ [LM]_{i,k} ; i \in V(G); k \in [0, 1, \dots, d(G)] \} \quad (4)$$

with  $d(G)$  standing for the diameter of the graph (*i.e.*, the largest distance in  $G$

## SHELL MATRICES

The entries in the shell matrix **ShM** (of a vertex pair property) are defined as follows [8]:

$$[ShM]_{i,k} = \sum_{v \mid d_{i,v}=k} [M]_{i,v} \quad (5)$$

The shell matrix is a collection of the above defined entries:

$$ShM(G) = \{ [ShM]_{i,k} ; i \in V(G); k \in [0, 1, \dots, d(G)] \} \quad (6)$$

A shell matrix **ShM**( $G$ ) will partition the entries of a square matrix according to the vertex (distance) partitions in the graph. It represents a true decomposition of the property collected by the info square matrix according to the contributions brought by vertex pairs pertaining to shells located at distance  $k$  around each vertex.

The shell matrices were recently used as the basis in the calculation of super-index super index Cluj-Niš<sup>3</sup> C/JN [3].

## SHELL-DEGREE-DISTANCE POLYNOMIALS

The Cramer product of the diagonal matrix of vertex degrees **D** with the distance **DI** matrix provides the matrix of degree distances [9] denoted **DDI**.

$$D(G) \times DI(G) = DDI(G) \quad (7)$$

The above Cramer product is equivalent (gives the same half sum of entries) with the pair-wise (Hadamard) product of the vectors “row sum” **RS** in the Adjacency **A** and Distance **DI** matrices, respectively.

$$RS(\mathbf{A}) \bullet RS(\mathbf{DI}) = RS(\mathbf{DDI}) \quad (8)$$

Next, by applying the Shell operator, we obtain the matrix **ShDDI**, of which column half sums are just the coefficients of the corresponding Shell-polynomial.

$$P(\mathbf{ShDDI}, x) = \sum_k p(G, k) \cdot x^k \quad (9)$$

An index, called *Cluj-Tehran*  $CT(\mathbf{ShM}, G)$ , with specified **M**, is defined as

$$CT(\mathbf{ShM}, G) = P'(\mathbf{ShM}, 1) + (1/2)P''(\mathbf{ShM}, 1) \quad (10)$$

where  $P'$  and  $P''$  refers to the polynomial first and second derivative, respectively (here calculate at  $x=1$ ) [9].

### ECCENTRIC CONNECTIVITY INDEX

Sharma *et al.* [10] introduced a distance-based molecular structure descriptor, the eccentric connectivity index, which is defined as:

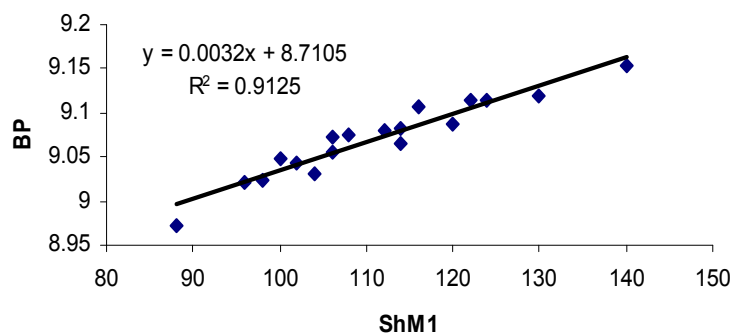
$$\xi^c(G) = \sum_{v \in V} \deg(v) \cdot \text{ecc}(v) \quad (11)$$

The eccentric connectivity index was successfully used for mathematical models of biological activities of diverse nature. The index  $\xi^c$  has been shown to give a high degree of predictability of pharmaceutical properties, and provide leads for the development of safe and potent anti-HIV compounds. The investigation of its mathematical properties started only recently, and has so far resulted in determining the extremal values and the extremal graphs, and also in a number of explicit formulas for the eccentric connectivity index of various products of graphs, several families of benzenoid graphs, zigzag and armchair hexagonal belts, nanotubes and nanotori [11].

### MODELING OCTANE PROPERTIES

To test the correlating ability of the descriptors derived from the degree-distance matrices (**ShDDI<sub>k</sub>**) and Shell-polynomials, we focused attention to the set of octanes, as one of the benchmark-sets [12,13] in correlating studies by using topological indices. Among several properties tested, three ones (listed in Table 1) provided best scores: boiling point BP, entropy S and total surface area TSA, in mono- to three- variable regressions, by our novel descriptors.


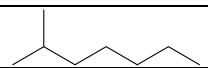
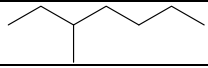
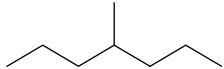
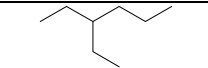
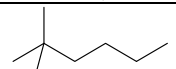
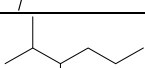

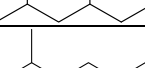
The octane topological descriptors are given in Table 2. They include the numbers derived from the  $\text{ShM}_1$  (the basic degree-distance matrix, in the Shell-matrix format), the first and second derivatives,  $D_1$  and  $DD_1$ , of the corresponding Shell-polynomial, at level  $k=1$  (equivalent to  $A_1$ ) and the same descriptors at  $k=2$  (equivalent to  $A_2$  – the exponent referring to the remote adjacency rank) and also Cluj-Tehran  $CT_1$  index and the eccentric connectivity index, denoted ECC in Table 2. Statistics are presented in Table 3.



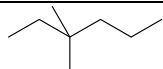
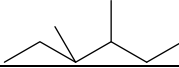

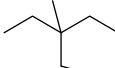
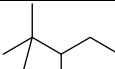
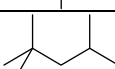
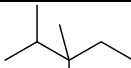
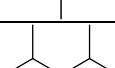

**Figure 1.** The plot of BP vs ShM<sub>1</sub> index

The best monovariate model of BP (by the index ShM<sub>1</sub>) is plotted in Figure 1. Compare our results with the best results reported so far in literature (Table 3, entry 7) and remark the best result ( $R^2 = 0.913$ ) for the monovariate model of the octane boiling points. The other results in Table 3 (mono- or tri- variable models) can be considered as good (or acceptable) results.

**Table 1.** Octanes boiling point BP, entropy S and total surface area TSA values

	Molecule	BP	S	TSA
1		9.153	111.67	415.3
2		9.120	109.84	407.85
3		9.115	111.26	397.34
4		9.114	109.32	396.04
5		9.108	109.43	379.04
6		9.065	103.42	405.11
7		9.079	108.02	384.93
8		9.082	106.98	388.11
9		9.088	105.72	395.08

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	Molecule	BP	S	TSA
10		9.056	104.74	389.79
11		9.074	106.59	376.91
12		9.073	106.06	368.1
13		9.049	101.48	366.99
14		9.023	101.31	371.75
15		9.031	104.09	392.19
16		9.020	102.06	377.4
17		9.044	102.39	368.93
18		8.971	93.06	390.47

**Table 2.** Topological Indices of Octanes

Molecule	ShM <sub>1</sub>	D <sub>1</sub>	DD <sub>1</sub>	CT <sub>1</sub>	ShM <sub>2</sub>	D <sub>2</sub>	DD <sub>2</sub>	ECC
1	140	532	1876	1470	258	972	3402	74
2	130	452	1406	1155	263	921	2900	65
3	124	404	1148	978	248	804	2270	63
4	122	388	1066	921	243	765	2070	61
5	116	340	808	744	230	668	1572	54
6	114	340	854	767	269	815	2088	56
7	112	320	746	693	244	704	1658	54
8	114	332	790	727	250	740	1796	54
9	120	376	998	875	266	856	2356	56
10	106	284	602	585	247	663	1390	52
11	108	292	620	602	234	636	1350	52
12	106	276	538	545	230	606	1202	43
13	100	244	432	460	232	568	996	41

Molecule	ShM <sub>1</sub>	D <sub>1</sub>	DD <sub>1</sub>	CT <sub>1</sub>	ShM <sub>2</sub>	D <sub>2</sub>	DD <sub>2</sub>	ECC
14	98	236	414	443	250	624	1138	43
15	104	272	546	545	268	732	1552	45
16	96	224	370	409	244	588	1000	41
17	102	256	476	494	243	633	1236	43
18	88	184	246	307	262	592	876	34

**Table 3.** Statistics ( $R^2$ ) of QSPR Study on Octanes

Descriptors		BP	S	TSA
1	ShM <sub>1</sub>	0.913	0.771	0.520
2	D <sub>1</sub>	0.863		
3	CT <sub>1</sub>	0.818	0.646	0.613
4	ECC	0.854	0.747	0.583
5	ShM <sub>2</sub> &ECC	0.938	0.873	0.902
6	ShM <sub>1</sub> &D <sub>1</sub> &DD <sub>1</sub>	0.987	0.933	0.914
7	ShM <sub>2</sub> &D <sub>2</sub> &DD <sub>2</sub>	0.991	0.924	0.870
8	Best in Octanes (monovariate)	(0.78) <sup>13</sup> (0.77) <sup>14</sup>	(0.92) <sup>13</sup> (0.93) <sup>14</sup> (0.950) <sup>1</sup>	(0.72) <sup>13</sup> (0.92) <sup>14</sup>

## CONCLUSIONS

A correlating study, using as independent variables the topological indices newly developed within the design of the super-index Cluj-Niș *CJN*, was performed on the set of octane isomers. The modeled properties were: boiling point BP, entropy S and total surface area TSA, which gave best scores, among several properties, in mono- to three-variable regressions, with respect to our novel descriptors. The most important result was the monovariate description of octane boiling points, which is the best result so far published in literature.

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